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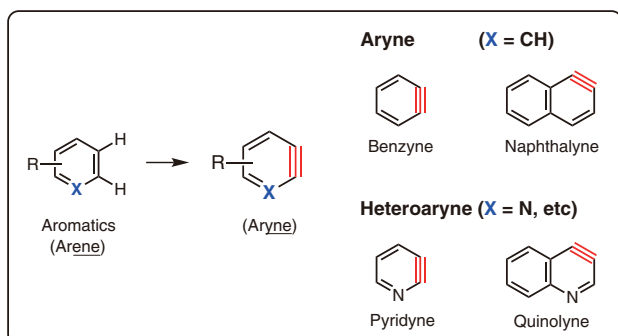
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Aryne / Heteroaryne Precursors

Arynes are the dehydro hydrocarbons derived from arenes, such as benzene or naphthalene, by abstraction of two hydrogen atoms from adjacent carbon atoms, and they characteristically have triple bonds (-yne) in their aromatic rings.



In addition, arynes which have hetero atoms, such as nitrogen, in their aromatic rings are called "**heteroarynes**". For examples of heteroarynes, pyridynes (derived from the pyridines) or quinolynes (derived from the quinolones) have been known so far.

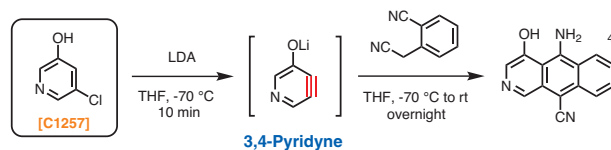
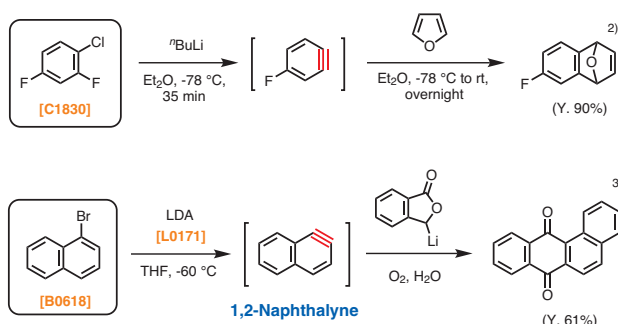
In general, arynes have extremely high reactivity due to their strained structures, thus, they have been widely used in organic synthesis as useful intermediates.¹⁾

● Synthetic methods

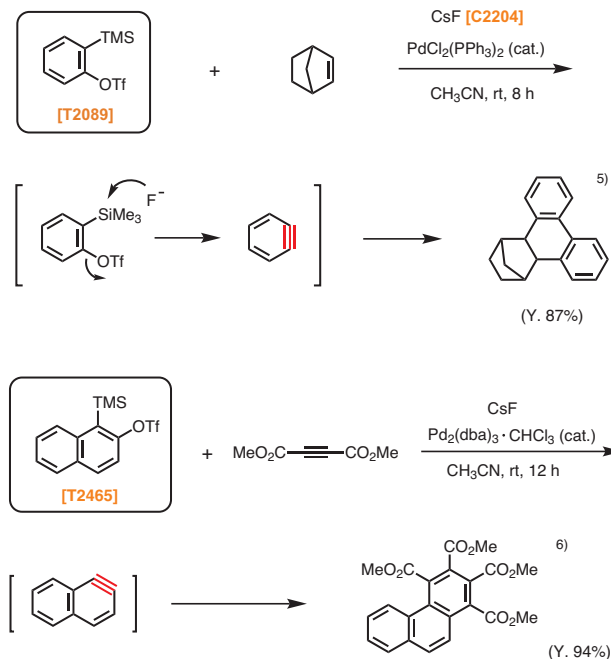
A number of methods for the generation of arynes have been reported so far. Arynes cannot be isolated because of their high reactivity, and they are generally prepared *in situ* in reaction systems. The typical examples are described as below.

1. The method using halogenated aryl compounds

Halogenated aryl compounds are treated with strong bases, such as ^tBuLi, NaNH₂ or Lithium Diisopropylamide (LDA) [L0171], to generate the corresponding arynes. The generation of arynes can be confirmed by trapping reactions with dienes, such as furan, affording the Diels-Alder cyclo-adducts.



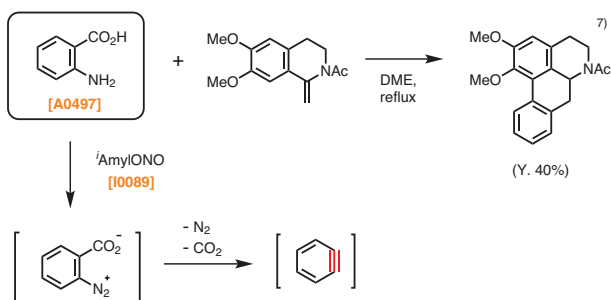
2. The method using 2-TMS-aryl triflates



The method using halogenated aryl compounds described in **section 1.** needs strong bases, such as ^tBuLi. For avoiding that, arynes precursors that can be used in milder conditions also have been developed. 2-TMS-aryl triflates react with fluoride salts, such as cesium fluoride [C2204], to generate the corresponding arynes under mild conditions.

3. The method using anthranilic acids

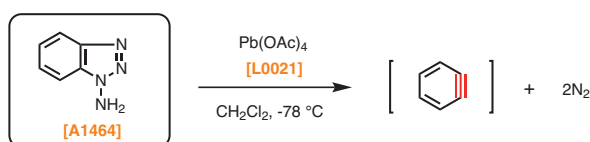
An alternative method using anthranilic acids has been reported. The diazo compounds derived from anthranilic acids are decomposed to generate the corresponding arynes, eliminating nitrogen and carbon dioxide. For the preparation of diazo compounds, nitrite salts are generally well-known. However, in recent years, amyl nitrite [I0089] or *tert*-butyl nitrite [N0357] are frequently used for simple use.



4. Other methods for aryne generation

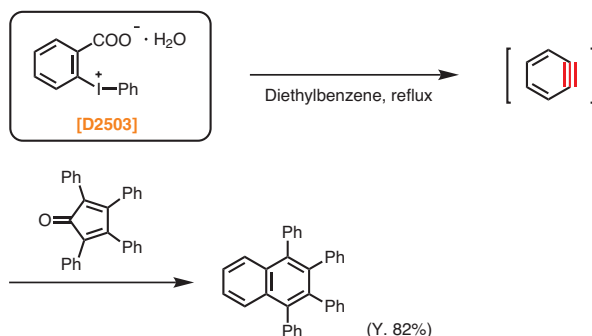
• Generation of benzyne from 1-aminobenzotriazole

Campbell *et al.* have reported the generation reaction of benzyne using 1-aminobenzotriazole [A1464], in which A1464 is oxidatively decomposed by lead acetate [L0021] to generate benzyne, eliminating nitrogen.⁸⁾

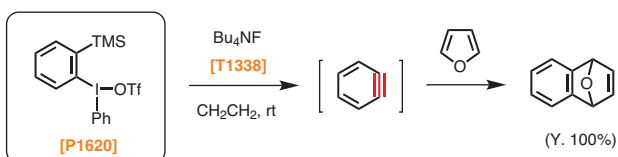


• The generation of benzyne using hypervalent iodine compounds

Diphenyliodonium-2-carboxylate [D2503] has been reported to generate benzyne under reflux conditions in diethylbenzene.⁹⁾



Moreover, phenyl[2-(trimethylsilyl)phenyl]iodonium trifluoromethanesulfonate [P1620], developed by Kitamura *et al.* is a mild benzyne precursor, which is treated with a fluoride salt to efficiently generate benzyne at room temperature.¹⁰⁾

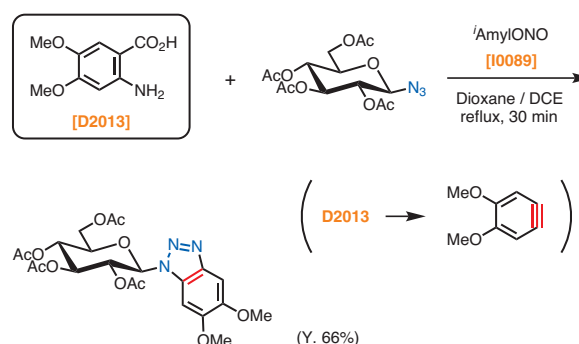


● Reaction examples of arynes

Arynes are widely used as many kinds of synthetic intermediates. The typical application examples are described as below.

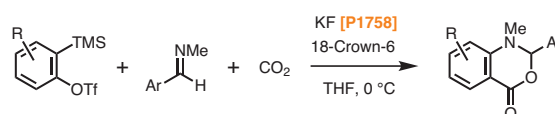
1. Click chemistry

Watt *et al.* have reported the synthesis of glucopyranose derivatives bearing a benzotriazolyl group *via* the Huisgen reaction (click reaction) of benzynes derived from the corresponding anthranilic acids and sugar azides, and their glycosyl donor abilities.¹¹⁾ In the Huisgen reaction, metal catalysts, such as copper sulfate, are generally required for reaction acceleration, however, the strain of the triple bond of the benzyne promotes the reaction without adding metal catalysts.



2. Multicomponent reaction

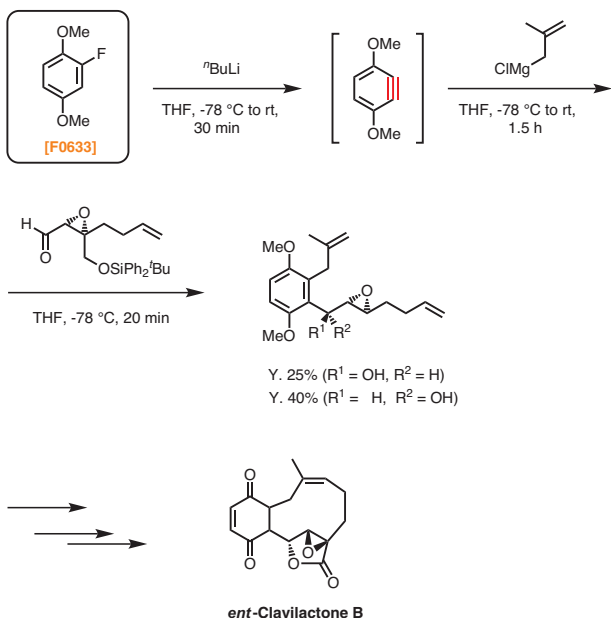
Arynes also have been used for multicomponent reactions (MCR). For example, Yoshida *et al.* have reported the three-component MCR using in situ generated benzynes, imines, and carbon dioxide, affording benzoxadionones.¹²⁾ Recently, much attention has been paid to organic synthesis using carbon dioxide as a carbon source from the ecological point of view, thus, the reaction above is an extremely useful and eco-friendly reaction.



Ar = 2,4,6-trimethylphenyl		
Benzyne Precursor	Reaction time (h)	Product
	15	
	46	
	60	

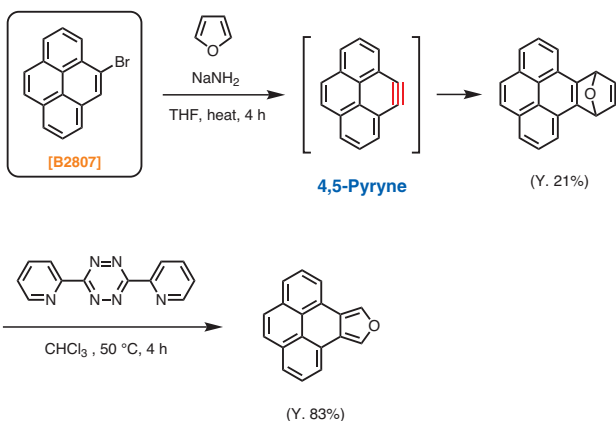
3. Synthesis of key intermediates for total synthesis (*ent*-Clavilactone B)

Arynes are also useful building blocks in total synthesis. For example, Barret *et al.* have reported the total synthesis of a natural product, *ent*-Clavilactone B, which shows tyrosine kinase inhibitory activity, using a benzyne derivative as a key starting material.¹³⁾



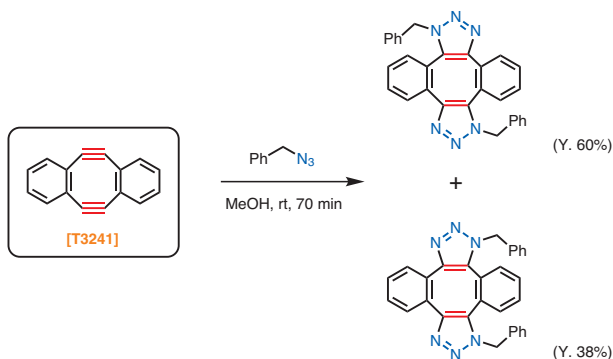
Other aryne precursors and related compounds

As described above, benzyne, naphthalene, or pyridine are relatively well-known as examples of arynes, however, several arynes derived from other aromatic compounds also have been reported. For example, Moursounidis *et al.* have reported the generation of 4,5-pyryne, derived from 4-bromopyrene **[B2807]**, and the reaction using 4,5-pyryne affording the pyrene-annulated furan derivatives.¹⁴⁾



In general, “cycloalkynes”, such as arynes, have a strained structure, thus, arynes cannot be isolated. On the other hand, 5,6,11,12-tetradehydridibenzo[a,e]cyclooctene **[T3241]**, reported by Sondheimer *et al.*, is relatively more stable than other arynes enough to isolate.¹⁵⁾ The compound also has a strained structure, and has been reported to have high reaction activity. Hosoya *et al.* have reported the “double-click reaction” applying **T3241** in click chemistry. The high reactivity of the two alkyne moieties allows the reaction to proceed smoothly without using metal catalysts, such as a copper salt.¹⁶⁾

Metal-Free Double-click Reaction



TCI offers a variety of aryne precursors. All the products in this brochure have been reported to generate the corresponding arynes so far, thus, reaction applications and related information for each product are introduced on our website. In addition, reagents for the generation of arynes are also listed in this brochure.

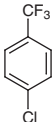
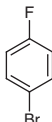
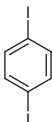
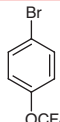
References

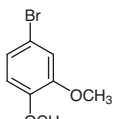
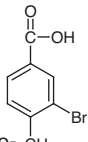
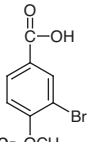
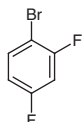
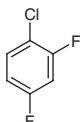
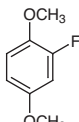
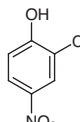
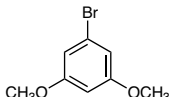
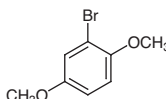
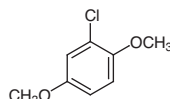
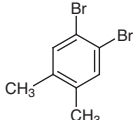
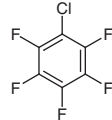
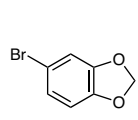
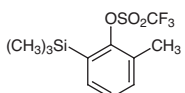
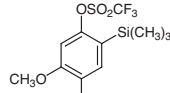
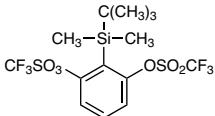
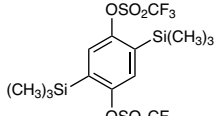
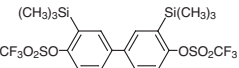
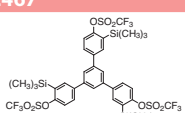
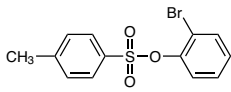
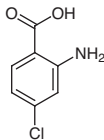
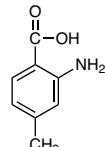
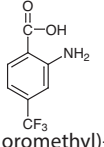
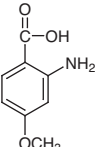
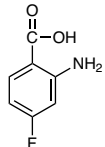
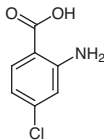
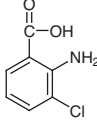
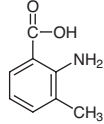
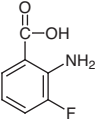
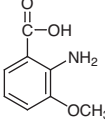
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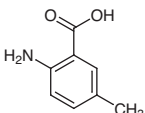
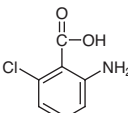
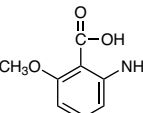
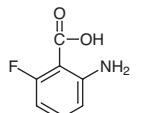
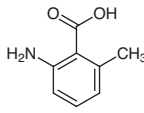
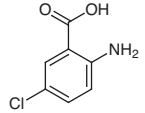
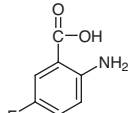
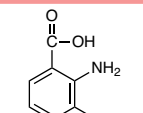
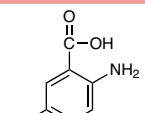
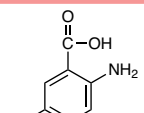
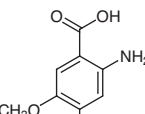
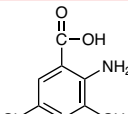
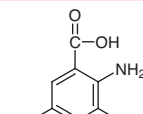
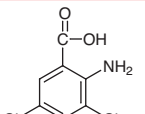
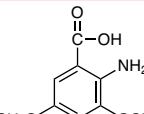
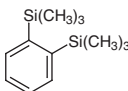
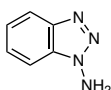
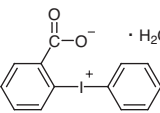
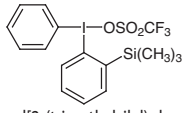
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Benzyne Precursors

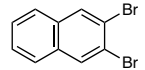
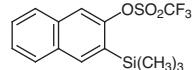
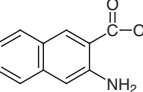
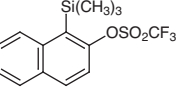
Halogenated Benzenes

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C0305 25g 500g  4-Chlorobenzotrifluoride CAS RN: 98-56-6	B0884 25g 100g 500g  4-Bromofluorobenzene CAS RN: 460-00-4
D0608 5g 25g  1,4-Diiodobenzene CAS RN: 624-38-4	B1772 5g 25g  1-Bromo-4-(trifluoromethoxy)-benzene CAS RN: 407-14-7
C0648 25g  1-Chloro-3-fluorobenzene CAS RN: 625-98-9	C1148 25g  3-Chloroanisole CAS RN: 2845-89-8
D0168 25g 100g  1,2-Dibromobenzene CAS RN: 583-53-9	B0663 25g 250g  2-Bromobenzotrifluoride CAS RN: 392-83-6
C0973 25g  2-Chlorobenzyl Alcohol CAS RN: 17849-38-6	C0647 10g 25g  2-Chlorofluorobenzene CAS RN: 348-51-6

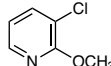
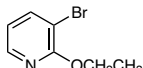
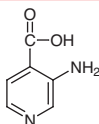
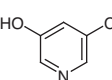
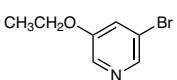
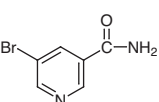
F0253 10g 25g  1-Fluoro-2-iodobenzene CAS RN: 348-52-7	B1021 25g  4-Bromo-1,2-dimethoxybenzene CAS RN: 2859-78-1	B3049 5g 25g  3-Bromo-4-methylbenzoic Acid CAS RN: 7697-26-9	B3336 5g 25g  3-Bromo-4-methoxybenzoic Acid CAS RN: 99-58-1	C0313 25g 500g  2-Chloro- <i>p</i> -xylene CAS RN: 95-72-7
D1909 25g 500g  1-Bromo-2,4-difluorobenzene CAS RN: 348-57-2	C1830 25g  1-Chloro-2,4-difluorobenzene CAS RN: 1435-44-5	F0633 5g  2-Fluoro-1,4-dimethoxybenzene CAS RN: 82830-49-7	C0227 25g 500g  2-Chloro-4-nitrophenol CAS RN: 619-08-9	B3848 5g 25g  1-Bromo-3,5-dimethoxybenzene CAS RN: 20469-65-2
B1979 5g 25g  1-Bromo-2,5-dimethoxybenzene CAS RN: 25245-34-5	C1577 25g 500g  1-Chloro-2,5-dimethoxybenzene CAS RN: 2100-42-7	D2272 5g 25g  1,2-Dibromo-4,5-dimethylbenzene CAS RN: 24932-48-7	P0850 25g  Chloropentafluorobenzene CAS RN: 344-07-0	B1230 5g 25g  4-Bromo-1,2-methylenedioxybenzene CAS RN: 2635-13-4
OTf / TMS-Benzenes				
M1883 1g 5g  2-Methyl-6-(trimethylsilyl)phenyl Trifluoromethanesulfonate CAS RN: 556812-44-3	D3883 1g 5g  4,5-Dimethoxy-2-(trimethylsilyl)phenyl Trifluoromethanesulfonate CAS RN: 866252-52-0	B5557 200mg 1g  2-(<i>tert</i> -Butyldimethylsilyl)-1,3-phenylene Triflate CAS RN: 1637638-66-4	B5559 1g  2,5-Bis(trimethylsilyl)-1,4-phenylene Triflate CAS RN: 613676-07-6	B3047 1g  3,3'-Bis(trimethylsilyl)biphenyl-4,4'-diyl Bis(trifluoromethanesulfonate) CAS RN: 828282-80-0
T2467 1g  1,3,5-Tris[4-(trifluoromethanesulfonyloxy)-3-(trimethylsilyl)phenyl]benzene CAS RN: 847925-63-7	OTf-Benzenes			B4528 1g 5g  2-Bromophenyl <i>p</i> -Toluenesulfonate CAS RN: 84672-48-0
OTs-Benzenes				
Anthranilic Acid Derivatives				
A0661 25g  4-Chloroanthranilic Acid CAS RN: 89-77-0	A2538 1g 5g  2-Amino- <i>p</i> -toluic Acid CAS RN: 2305-36-4	A2175 5g 25g  4-(Trifluoromethyl)-anthranilic Acid CAS RN: 402-13-1	A2319 1g 5g  2-Amino- <i>p</i> -anisic Acid CAS RN: 4294-95-5	F0405 1g 5g 25g  4-Fluoroanthranilic Acid CAS RN: 446-32-2
A0661 25g  4-Chloroanthranilic Acid CAS RN: 89-77-0	A0786 5g 25g  3-Chloroanthranilic Acid CAS RN: 6388-47-2	A1569 5g 25g  2-Amino- <i>m</i> -toluic Acid CAS RN: 4389-45-1	F0570 1g 5g  3-Fluoroanthranilic Acid CAS RN: 825-22-9	A1378 5g 25g  2-Amino-3-methoxybenzoic Acid CAS RN: 3177-80-8

A1421 5g 25g  6-Amino- <i>m</i> -toluic Acid CAS RN: 2941-78-8	C2048 5g 25g  6-Chloroanthranilic Acid CAS RN: 2148-56-3	A2847 1g 5g  6-Amino- <i>o</i> -anisic Acid CAS RN: 53600-33-2	F0475 1g 5g  6-Fluoroanthranilic Acid CAS RN: 434-76-4	A0996 5g 25g  2-Amino-6-methylbenzoic Acid CAS RN: 4389-50-8
A0665 5g 25g  5-Chloroanthranilic Acid CAS RN: 635-21-2	F0396 1g 5g  5-Fluoroanthranilic Acid CAS RN: 446-08-2	D4120 5g 25g  3,4-Dimethylantranilic Acid CAS RN: 50419-58-4	A2850 200mg 1g  4,5-Dimethylantranilic Acid CAS RN: 15089-51-7	D4063 5g 25g  4,5-Difluoroanthranilic Acid CAS RN: 83506-93-8
D2013 10g  4,5-Dimethoxyanthranilic Acid CAS RN: 5653-40-7	A2399 5g 25g  2-Amino-5-chloro- <i>m</i> -toluic Acid CAS RN: 20776-67-4	D2553 1g  3,5-Dimethylantranilic Acid CAS RN: 14438-32-5	D1475 5g 25g  3,5-Dichloroanthranilic Acid CAS RN: 2789-92-6	A2770 1g  3,4,5-Trimethoxyanthranilic Acid CAS RN: 61948-85-4
Other Benzyne Precursors	B2299 1g 5g  1,2-Bis(trimethylsilyl)benzene CAS RN: 17151-09-6	A1464 1g 5g  1-Aminobenzotriazole CAS RN: 1614-12-6	D2503 5g 25g  Diphenyliodonium-2-carboxylate Monohydrate CAS RN: 96195-89-0	P1620 1g 5g  Phenyl[2-(trimethylsilyl)phenyl]iodonium Trifluoromethanesulfonate CAS RN: 164594-13-2

Naphthalene Precursors

D4597 200mg 1g  2,3-Dibromonaphthalene CAS RN: 13214-70-5	T2466 1g 5g  3-(Trimethylsilyl)-2-naphthyl Trifluoromethanesulfonate CAS RN: 780820-43-1	A2258 1g 5g  3-Amino-2-naphthoic Acid CAS RN: 5959-52-4	T2465 1g 5g  1-(Trimethylsilyl)-2-naphthyl Trifluoromethanesulfonate CAS RN: 252054-88-9	C2310 5g 25g  1-Chloronaphthalene CAS RN: 90-13-1	B0618 25g 100g 500g  1-Bromonaphthalene CAS RN: 90-11-9	F0212 5g 25g  1-Fluoronaphthalene CAS RN: 321-38-0
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Pyridine Precursors

Pyridyne Precursors		B0651 25g 100g  3-Bromopyridine CAS RN: 626-55-1	C0280 25g 100g 500g  3-Chloropyridine CAS RN: 626-60-8	C2565 1g 5g  3-Chloro-2-methoxypyridine CAS RN: 13472-84-9
B4738 1g  3-Bromo-2-ethoxypyridine CAS RN: 57883-25-7	A2133 1g 5g  3-Aminoisonicotinic Acid CAS RN: 7579-20-6	C1257 5g  3-Chloro-5-hydroxypyridine CAS RN: 74115-12-1	B5000 1g 5g  3-Bromo-5-ethoxypyridine CAS RN: 17117-17-8	B3536 5g  5-Bromonicotinamide CAS RN: 28733-43-9

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