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Organic Transistor (OFET) Materials

Organic field-effect transistors (OFETs) are promising components for the next-generation electronic devices. In 1984, OFET research began with the first report of mobility in a merocyanine dye-based field-effect device by Kudo *et al.*¹⁾ Since high hole mobility (1.5 cm²/Vs), which comparable to that of amorphous silicon, was achieved using a pentacene-based OFET device in 1997,²⁾ the possibility of a practical application of OFET's became realistic and the research field became quite popular worldwide. While silicon has performed well in devices, the inorganic nature of them prohibits flexible structures. As a result, OFETs have attracted much attention for their potential to be flexible, thin and light-weight, which could be applicable for foldable electronic circuits and implantable biometric sensors. ^{3, 4)}

A noted potential application with OFET's involves their "printability". Printed electronics are an innovative technology for mass and low cost device productions, which could allow for the production of high density and large circuits on flexible substrates such as paper and plastic films. A fusion of "Printed electronics" and "Organic transistors" could offer especially promising technology by allowing for low cost and large scale manufacture of various functional devices. ⁵⁻⁷⁾

One of the functional parameters evaluated in organic semiconductor materials is mobility (μ), which indicates how fast the holes (p-type) or electrons (n-type) within the semiconducting layer move. A material that possesses high carrier mobility is required for producing high-speed circuits. OFET devices are a largely simple construction containing an organic semiconductor layer, an insulating layer, and source-drain-gate electrodes. These components allow for the evaluation of fundamental transistor parameters including mobility, operation voltage, and driving stability.



Due to the expanded π -conjugated system within organic semiconductor molecules, these often produce large intermolecular interactions inducing an improvement in the mobility within OFET

devices. In general, the expansion and extension of π -conjugation is an effective strategy in the molecular design of OFET molecules and polymers. As an example, Pentacene, which consists of 5 fused benzene rings, possesses superior electrical properties compared to that of Tetracene, which consists of 4 fused benzene rings.⁸⁾ However, a linear increase in the number of fused aromatic rings in a simple hydrocarbon system raises the Highest Occupied Molecular Orbital (HOMO), resulting in a critical decrease in the air-stability of the molecule and semiconductor materials.⁸⁾ This trade-off issue between the mobility and the air-stability of the materials has been a key problem to overcome. Given this context, in 2006 DPh-BTBT [D3526], a thienothiophene-fused organic compound, had been reported by Takimiya et al., as an innovative OFET material.⁹⁾ DPh-BTBT features a deep HOMO level (-5.6) eV), which leads to remarkable air-stability during OFET device performance, and the HOMOs are well-distributed over the sulfur atoms of the thienothiophene moieties inducing good hole carrier transport. DPh-BTBT-based FET devise further achieved excellent electrical properties with a high hole mobility of 2.0 cm²/Vs. Followed by the innovation and molecular design of DPh-BTBT, Ph-BTBT-10, an smectic E (SmE) liquid crystalline material, was recently reported by Hanna et al., as a p-type material that bears an asymmetric structure in which the alkyl and the phenyl group are substituted on one side of the benzothienobenzothiophene (BTBT) moiety.¹⁰⁾ Additionally, Ph-BTBT-10 can be handled in solution processing such the spin-coting method, and possesses both good heat-resistance and film-forming properties. The Ph-BTBT-10 spin-coated devise exhibited not only outstanding OFET performance with ultra-high mobility (μ_{max} = 14.7 cm²/Vs), but also comparable to oxide semiconductors (IGZO) with remarkable airstability.



One of the largest benefits of organic compounds for OFET's is their immense structural variety. This variety can be used to control the various properties they possess such as the electrical performance, stability, and processing characteristics through the chemical modification of the molecular units. Within the organic transistor research field, the possibilities for practical application has dramatically improved by introducing new concepts and various materials proposals, including the unique OFET materials and the parent molecules used to create them.

High Quality Organic Semiconductor Materials (For Organic Electronics)

TCI offers "High Quality Organic Semiconductor Materials (For Organic Electronics)" specialized for electrical performance such as OFET mobility*. A material used in the active layer of an OFET device requires exceptionally high-purity to produce good OFET function. However, it is difficult to analytically (HPLC, GC, etc.) determine the purity at the ultra-high levels required for OFET function. To surmount this quality assurance challenge, we have begun in-house fabrication of OFET devices using our OFET materials. Once fabricated, we assess the functionality of the OFET as a quality assurance measure to confirm the electronic properties and device performance of the "High Quality Organic Semiconductor Materials (For Organic Electronics)". We constantly seek to improve our technology and skill in order to provide high-purity and quality materials our customers.

(*The mobility refers to device evaluation measurements obtained within our facility under environment condition.)



 List of High Quality Organic Semiconductor Materials (For Organic Electronics)



		Specification			
Compound	Product No.	Purity (%)	Mobility (cm²/Vs)	Si/SiO ₂ Substrate Surface Condition	
Pentacene	P2524	> 99.999	> 0.35	Bare	
Ph-BTBT-10	D5491	> 99.5	> 10.0	ODTS	
P3HT	P2513	Pd: < 100ppm	> 0.10	OTS	
TIPS Pentacene	B5942	> 99.0	> 0.10	HMDS	
C ₇₀	F1233	> 99.0	> 0.30	HMDS	
[60]PCBM	P2682	> 99.5	> 0.020	HMDS	
[70]PCBM	P2683	> 99.0	> 0.015	HMDS	

1. An Example of OFET Evaluation 1: Typical p-type Material "Pentacene"

Pentacene

(99.999%, trace metals basis) (purified by sublimation)

[P2524]



Pentacene, a simple polyaromatichydrocarbon, has been studied for its fundamental properties and applications in organic electronic research.^{11, 12)} In particular, many research studies have been conducted to evaluate its potential as a carrier transporter within OFET devices.^{2,13)} TCI provides two subliminally purified pentacene reagents: **P2524** and **P0030**. **P2524** (99.999% trace metal basis) is our high-quality grade reagent, which has additional specifications for its OFET mobility: [> 0.35 cm²/Vs (bare Si/SiO₂)]. We inspect **P2524** through the OFET evaluation process in every product lot. Only lots that pass this functional test are packed and shipped as "High Quality Organic Semiconductor Materials (For Organic Electronics)".



Figure 1. Transfer curves of various grades of pentacene.

Table 1. OFET characteristics of the pentacenes.

Product No. Sample grade		Substrate	Mobility (cm²/Vs)	V _{th} (V)
-	Pentacene (non-sublimation)	Si/SiO ₂ (bare)	5.3 × 10 ⁻⁴	-13
P0030	Pentacene (purified by sublimation)	Si/SiO ₂ (bare)	0.29	-22
P2524	Pentacene (99.999%, trace metals basis) (purified by sublimation)	Si/SiO ₂ (bare)	0.39	-10

The performances of the pentacene-based OFET devices are summarized in Table 1 and Figure 1. These devices were fabricated via vacuum deposition method on bare Si/SiO₂ substrate without Self-Assembled Monolayer (SAM) treatment, the characteristics of which were measured under nitrogen conditions. Our sublimed pentacene [P2524] showed a large increase OFET performance compared to that of the non-sublimed pentacene. As a result, P2524 (99.999% trace metal basis) showed an excellent OFET performance with the highest hole mobility of 0.39 cm²/Vs. In comparison to other companies' sublimed pentacene samples, P2524 (99.999% trace metal basis) possesses the highest drain current and the best OFET potential (Figure 2 and Table 2).



Figure 2. Transfer curves of OFET devices using several companies' pentacene.

Table 2. OFET characteristics of the pentacenes.

	Substrate	Mobility (cm²/Vs)	V _{th} (V)
P2524	Si/SiO ₂ (bare)	0.39	-10
Company A (sublimed)	Si/SiO ₂ (bare)	0.002	-25
Company B (sublimed)	Si/SiO ₂ (bare)	0.001	-25
Company C (sublimed)	Si/SiO ₂ (bare)	5.0 × 10 ⁻⁶	-23

1-a. Optimization of pentacene-based OFET devices

We have examined and optimized the functional preformance of pentacene-based OFET devices via substrate surface modification. The field-effect mobility of pentacene was measured using top-contact thin-film field-effect transistors geometry (Figure 3). The thin film of pentacene [P2524] as the active layer (60 nm) was vacuum-deposited onto Si/SiO₂ substrate (bare) or *n*-Octyltrichlorosilane (OTS) [O0168] -treated Si/SiO₂ substrate at room temperature ($T_{sub} = RT$). The drain and source electrodes (40 nm) were then prepared by gold evaporation through a shadow mask on top of the pentacene film; the drain-source channel length (*L*) and width (*w*) were 50 µm and 1.5 mm, respectively. The characteristics of the OFET devices were measured under nitrogen conditions.

The performances of the OFET devices are summarized in Table 3 and figure 4. All pentacene -based devices exhibited pure typical p-channel field-effect transistor (FET) characteristics. The FET performance were significantly improved by the SAM treatment; the OTS-treated device demonstrated the highest performance with a hole carrier mobility of 1.52 cm²/Vs and an on/off ratio of 1.5 $\times 10^7$ (Figure 4).

Figure 3. Illustration for the device structure of pentacene-based OFET device.



-igure 4. Typical OFE1 characteristics of top-contact devices fabricated using pentacene [P2524].
(a, c) bare. (b, d) OTS-treated substrate.
(a, b) Transfer curves in the saturated region.
(c, d) Output curves at different gate voltages.

Table 3. UFET characteristics of P2524-based devic	tics of P2524-based devices.
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Compound	SAM	T _{sub} (°C)	Mobility (cm²/Vs)	V _{th} (V)	on/off
Pentacene	bare	RT	0.35 - 0.37	-5.3	3.9 × 105
[P2513]	OTS [00168]	RT	1.50 - 1.52	-5.7	1.5 × 10 ⁷

1-b. AFM images and XRD analysis

To clarify the pentacene thin film morphology and conformation, atomic force microscope (AFM) and X-ray diffraction (XRD) measurement were carried out (Figure 5). Regardless of either bare or OTS-treated substrate, highly regular terrace structures were observed (Figure 5a). In addition, XRD measurements of the pentacene films showed a series of peaks assignable to (00h) reflections; and the diffraction peak at $2\theta = 5.72^{\circ}$ corresponds to a *d*-spacing of 15.5 Å (Figure 5b). These results demonstrated that the pentacene molecules stood nearly perpendicular to the substrate (thin-film-phase) in the film form. ^{12, 13)}



Figure 5. AFM images(a). and XRD analysis(b) of pentacene filims.

In the bare device (without SAM), two weak peaks assignable to face-on orientation were observed (Figure 5b, black arrow), which would create the disadvantage of carrier passes parallel to the substrate (Figure 6a). In contrast, such peaks were not observed in the pentacene film on the OTS-treated substrate (Figure 6b). It is one reason why the mobility was higher in the OTS-treated OFET device.



Figure 6. Orientation images of pentacene thin film form.

2. An Ultra-high Performance p-type Semiconductor Material "Ph-BTBT-10"

Ph-BTBT-10 (= 2-Decyl-7-phenyl[1]benzothieno[3,2-b][1]benzothiophene) [D5491]



[D5491]

Ph-BTBT-10, an SmE liquid crystalline material, has been recently reported by Hanna *et al.*, as a p-type material which possesses excellent transport properties.¹⁰⁾ Ph-BTBT-10 exhibits ultra-high mobility (μ_{max} = 14.7 cm²/Vs) comparable to oxide semiconductors (IGZO), and remarkable air stability. TCI has recently commercialized Ph-BTBT-10 as an ultra-high mobility and air stability p-type OFET material, and has begun evaluation via the fabrication and performance measurement of Ph-BTBT-10-based OFET devices using vacuum deposition methods in our laboratories. The devise showed a hole carrier mobility up to 14.0 cm²/Vs. Please see below for details.

2-a. Fabrication and evaluation of Ph-BTBT-10-based OFET device

The field-effect mobility of Ph-BTBT-10 was measured using topcontact thin-film field-effect transistors geometry (Figure 7). The thin film of Ph-BTBT-10 as the active layer (40 nm) was vacuumdeposited onto Si/SiO₂ substrate (bare) or Octadecyltrichlorosilane (ODTS) **[O0079]**-treated Si/SiO₂ substrate while heating the substrate. The drain and source electrodes (40 nm) were then prepared by gold evaporation through a shadow mask on top of the Ph-BTBT-10 film; the drain-source channel length (*L*) and width (*w*) are 50 µm and 1.5 mm, respectively. After deposition, these devices were thermal annealed at T_{sub} = 120 °C for 5 min under ambient conditions, and the characteristics of the OFET devices were measured.



Figure 7. Illustration for the device structure of Ph-BTBT-10-based OFET.

The performances of the OFET devices are summarized in Table 4 and Figure 8. All Ph-BTBT-10-based devices exhibited pure typical p-channel field-effect transistor (FET) characteristics. The FET mobilities were quite dependent on the thermal annealing treatment regardless of the self-assemble-monolayer (SAM) (Figure 8). This would be attributed to the phase transition from a monolayer to a bilayer crystal structure in the thin-film form. ¹⁰⁾ The device fabricated on bare substrate demonstrated good performance with a hole carrier mobility of 4.86 cm²/Vs and threshold voltage (V_{th}) of -8 V. Moreover, although V_{th} increased, the ODTS-treated devise showed the highest transport performance with a hole carrier mobility of 14.0 cm²/Vs.

Table 4. OFET characteristics of the Ph-BTBT-10-based devices

Compound	SAM	Annealing Temp. (°C)	Mobility (cm²/Vs)	V _{th} (V)
Ph-BTBT-10	bare	w/o 120	0.87 - 0.91 4.24 - 4.86	-24 -8
[D5491]	ODTS [00079]	w/o 120	1.40 - 1.42 10.3 - 14.0	-23 -22



-20 V_G / V

0

0

-60

-40



Figure 8. Transfer curves of the Ph-BTBT-10-based OFET devices. (a) w/o annealing (bare) (b) annealing 120°C, 5min (bare) (c) w/o annealing (ODTS) (d) annealing 120°C, 5min (ODTS)

2-b. 2D-GIXD Analysis



Figure 9. (a) 2D-GIXD analysis (Thermal in situ mesurement), (b) 2D-GIXD analysis (Out-of-plane, bare substrate), (c) Phase transition images of Ph-BTBT-10

1x10⁻¹²

20

The crystal structures of Ph-BTBT-10 thin films were analyzed by 2D-GIXD using synchrotron radiation. Figure 9 (a) and (b) display the thermal in situ 2D-GIXD data at different temperatures. The Ph-BTBT-10 films at RT and 60°C showed a similar series of peaks assignable to a monolayer structure with a d-spacing of 27 Å. When the substrate temperature was increased to 120°C, the diffraction peaks clearly changed, which suggested a transformation from a monolayer structure (d = 27 Å) into a bilayer structure with a *d*-spacing of 54 Å as shown in Figure 9 (c). Since the phase transition temperature to the SmE mesophase is 144°C, $^{\scriptscriptstyle 10)}$ a series of peaks assignable to SmE were observed when heated to 180°C. In addition, a mixed layer of monolayer and bilayer structures appeared when the Ph-BTBT-10 thin film was rapidly cooled from 180°C to RT. These results indicate that under these conditions the cooling speed might be a significant factor in forming a well-uniformed crystalline thin film structure. Based on these results, Ph-BTBT-10 can be handled through vacuum deposition method, and the phase transition from the monolaver to the bilayer structure can occur in the same way in which it occurs for solution processed Ph-BTBT-10 thin films. ¹⁰⁾ Finally, we demonstrated top-ranked FET performances via vacuum deposition process using our in-house equipment.

3. Performance Evaluation of Highly Regioregular "P3HT"

P3HT (= Poly(3-hexylthiophene-2,5-diyl)) (regioregular) [P2513]



TCI's P3HT (Poly(3-hexylthiophene-2,5-diyl)) **[P2513]** features a very high regioregularity (RR > 99%), a narrow molecular weight distribution (Mn = $36k \sim 45k$), and a low metal content ratio (Pd < 100 ppm) in order to provide high quality solution-processed organic materials for organic electronics. The synthesis was conducted via the direct aryration polymerization (DArP) method in collaboration with Prof. Fumiyuki Ozawa at Institute for Chemical Research, Kyoto University.^{14, 15)}

We fabricated organic thin-film transistor (OFET) devices to validate and demonstrate hole transport properties of P3HTs. By comparing OFET characteristics of six P3HT samples including P2513, we revealed a correlation between OFET performances and (1) molecular weights and (2) regioregularities of P3HTs. As the result, the P2513 devise showed the highest hole mobility, among the other P3HT devices. This suggested that the carrier transport property of P3HT was improved by both the regioregularity and the molecular weight of polymer parameters. In particular, the regioregularity could be considered the most significant factor for

enhancing the electrical properties of P3HT in an OFET devise.

3-a. Fabrication and Evaluation of P3HT-based OFET devices

The hole mobility of P2513 was measured using top-contact OFET geometry (Figure 10). The P3HT [P2513] was dissolved in chloroform:trichlorobenzene at a concentration of 10 mg/ml. The solution of P2513 was spin-coated (1500 RPM) onto *n*-Octyltrichlorosilane (OTS) [O0168] -treated Si/SiO₂ substrate in a nitrogen glove box, then thermally annealed for 30 min. A gold layer with 40 nm thickness was deposited in vacuum chamber to serve as drain and source electrodes through a shadow mask on top of the P2513 film; the drain-source channel length (*L*) and width (*W*) are 50 µm and 1.5 mm, respectively. The characteristics of the OFET devices were measured under nitrogen conditions. The other five P3HTs were also evaluated under the same protocol.



Figure 10. Illustration for the device structure of P3HT-based OFET.

3-b. A correlation between the hole transport mobility and the molecular weight18)

The device performances of **P2513** (high molecular weight Mn: 40K) and the other three P3HT samples having low molecular weights (Mn: 8K~28K), are summarized in Figure 11 and Table 5. Samples 1 and 2 were synthesized via the same DArP method same as with **P2513**.

The OFET performances of P3HT-based devices were improved by increasing the molecular weights of P3HT. In the case of P2513, the device achieved the highest transport performance with a hole mobility of 0.1 cm²/Vs and an on/off ratio of 9×10^4 . The high molecular weight could enhance a crystallinity of P3HT in the film form, that could be a reason why P2513 possesses the excellent performances of OFET. The molecular weight of P2513 was set 30K - 45K as a specification.

Table 5. Properties of P3HTs and mobility of OFET devices.

	RR (%)	M _n	Mobility (x 10 ⁻² cm ² /Vs)
P2513	99	40K	10.5±0.4
Sample 1	99	18K	5.9±0.3
Sample 2	99	8K	3.0±0.4
Sample 3	98	28K	6.5±0.7
Sample 4	91	39K	1.2±0.1
Sample 5	93	-	1.7±0.3



Figure 11. (a) Transfer curves of P3HT-based devices differ from molecular weight of P3HT, (b) A correlation between the hole transport mobility and the molecular weight.

3-c. A correlation between the hole transport mobility and the regioregularity¹⁹⁾

The device performances of **P2513** (the high regioregularity >99%) and the other two P3HT samples having low regioregularities (91, 93%) are summarized in Figure 12 and Table 5. Samples 4 and 5 indicate other companies' P3HT. The OFET performances of P3HT-based devices were drastically improved with increasing the regioregularities within P3HTs. While sample 4 (RR: 91%, Mn: 39K) displayed a high molecular weight as **P2513** (Mn: 40K), the hole transport mobility of the sample 4-based device was lower than that of the sample 2 (RR: 99%, Mn: 8K) -based device. From these results, the OFET performances of P3HT could be enhanced by the regioregularity rather than the molecular weight of P3HT.





Figure 12. (A) Transfer curves of P3HT-based devices differ from regioregularity of P3HT, (b) A correlation between the hole transport mobility and the regioregularity.

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Fundamental properties of Organic Transistor Materials



UV-vis absorption spectra (using in-house equipment)







Energy levels of HOMO



Energy levels of LUMO



Product No.	Compound	LUMO (eV)	HOMO (eV)	<i>E</i> g (eV)	Reference
N0951	Tetracene		5.1		20)
P2524	Pentacene		4.9		20)
T2233	Rubrene		5.3		20)
D4469	6,13-Diphenylpentace	ene 3.2	5.2	2.0	21)
B5942	TIPS-Pentacene	3.4	5.1	1.8	22)
P2207	Picene		5.6		23)
D3526	DPh-BTBT	2.4	5.6	3.2	24)
D5491	Ph-BTBT-10		5.7		25)
D4617	anti-DMADT	3.1	5.2		26)
D4618	<i>syn</i> -DMADT	3.0	5.2		26)
D5086	DBTTT	2.4	5.3	2.9	27)
S0220	6P	1.8	5.6	3.8	28)
S0504	6T	3.1	5.3	2.2	29)
Q0078	4T		5.3		30)
D4842	C6-4T		5.1	2.7	31)
D4877	C8-4T		5.1	2.7	31)
D4888	C10-4T	2.6	5.4		32)
B3441	BP1T		5.7		33)
P2513	P3HT		4.8		34)
P1628	CuPc	2.7	4.8	2.2	35)
P1795	H ₂ PC	2.7	5.0	2.2	35)
Z0037	ZnPc		5.0		36)
B1200	BEDT-TTF		4.7		37)
B4576	Bis-PCBM	3.6	6.0	2.4	38)
10900	ICBA	3.5	6.0	2.5	38)
B1641	C ₆₀	4.0	6.4	2.4	38)
F1233	C ₇₀	4.0	6.4		38)
P2682	[60]PCBM	3.8	6.2		38)
P2683	[70]PCBM	3.8	5.9		38)
C2415	C ₆₀ MC ₁₂	3.6			39)
D4429	PTCDI-C1	4.0	6.6	2.6	40)
H1194	F ₁₆ CuPc	4.5	6.3	1.8	40)
D4175	PTCDI-C ₈	3.9	6.3	2.4	41)
B4268	Perylene Orange	3.9	6.2	2.3	42)
T3034	TCNQ	4.2			43)
T1131	F ₄ -TCNQ	5.2			44)

References

- 1) K. Kudo, M. Yamashinaand, T. Moriizumi, Jpn. J. Appl. Phys. 1984, 23, 130.
- Y. Y. Lin, D. J. Gundlach, S. F. Nelson and T. N. Jackson, *IEEE Trans. Electron Devices* 1997, 44, 1325.
- M. Kaltenbrunner, T. Sekitani, J. Reeder, T. Yokota, K. Kuribara, T. Tokuhara, M. Drack, R. Schwödiauer, I. Graz, S. Bauer-Gogonea, S. Bauer, T. Someya, *Nature* 2013, 499, 458.
- 4) B. Chu, W. Burnett, JW. Chung, Z. Bao, Nature 2017, 355, 328.
- A. C. Arias, J. D. MacKenzie, I. McCulloch, J. Rivnay, A. Salleo, *Chem. Rev.* 2010, 110, 3.
- 6) B. Kang, W. H. Lee, K. Cho, ACS Appl. Mater. Interfaces 2013, 5, 2302.
- 7) K. Fukuda, T. Someya, Adv. Mater. 2017, 29, 1602736.
- J. G. Mei, Y. Diao, A. L. Appleton, L. Fang, Z. N. Bao, J. Am. Chem. Soc. 2013, 135, 6724.
- K. Takimiya, H. Ebata, K. Sakamoto, T. Izawa, T. Otsubo, Y. Kunugi, J. Am. Chem. Soc. 2006, 128, 12604.
- 10) H. lino, T. Usui, J.-I. Hanna, Nat. Commun. 2015, 6, 6828.
- R. Ruiz, D. Choudhary, B. Nickel, T. Toccoli, K.-C. Chang, A. C. Mayer, P. Clancy, J. M. Blakely, R. L. Headrick, S. lannotta, G. G. Malliaras, *Chem. Mater.* 2004, 16, 4497.
- 12) B. Nickel, M. Fiebig, S. Schiefer, M. Göllner, M. Huth, C. Erlen, P. Lugli, *Phys. Stat. Sol. A* **2008**, 205, 526
- 13) M. Kitamura, Y. Arakawa, J. Phys.: Condens. Matter. 2008, 20, 184011
- 14) Q. Wang, R. Takita, Y. Kikuzaki, F. Ozawa, J. Am. Chem. Soc. 2010, 132, 11420.
- 15) J.-R. Pouliot, M. Wakioka, F. Ozawa, Y. Li, M. Leclerc, *Macromol. Chem. Phys.* 2016, 217, 1493.
- 16) R. Zhang, B. Li, M. C. Iovu, M. Jeffries-EL, G. Sauvé, J. Cooper, S. Jia, S. Tristram-Nagle, D. M. Smilgies, D. N. Lambeth, R. D. McCullough, T. Kowalewski, J. Am. Chem. Soc. 2006, 128, 3480.
- H. Sirringhaus, P. J. Brown, R. H. Friend, M. M. Nielsen, K. Bechgaard, B. M. W. Langeveld-Voss, A. J. H. Spiering, R. A. J. Janssen, E. W. Meijer, P. Herwig, D. M. de Leeuw, *Nature* **1999**, *401*, 685.
- 18) J.-F. Chang, B. Sun, D. W. Breiby, M. M. Nielsen, T. I. Sölling, M. Giles, I. McCulloch, H. Sirringhaus, *Chem. Mater.* 2004, *16*, 4772.
- L. A. Majewski, J. W. Kingsley, C. Balocco, A. M. Songa, *Appl. Phys. Lett.* 2006, 88, 222108.
- 20) N. Sato, K. Seki, H Inokuchi, J. Chem. Soc., Faraday Trans. 2 1981, 77, 1621.
- 21) L. C. Picciolo, H. Murata, Z. H. Kafafi, Appl. Phys. Lett. 2001, 78, 2378.

- 22) I. Kaur, W. Jia, R. P. Kopreski, S. Selvarasah, M. R. Dokmeci, C. Pramanik, N. E. McGruer, G. P. Miller, J. Am. Chem. Soc. 2008, 130, 16274.
- 23) T. Hosokai, A. Hinderhofer, F. Bussolotti, K. Yonezawa, C. Lorch, A. Vorobiev, Y. Hasegawa, Y. Yamada, Y. Kubozoro, A. Gerlach, S. Kera, F. Schreiber, N. Ueno, J. Phys. Chem. C 2015, 119, 2902.
- 24) K. Takimiya, H. Ebata, K. Sakamoto, T. Izawa, T. Otsubo, Y. Kunugi, *J. Am. Chem. Soc.* **2006**, *128*, 12604.
- 25) M. Kunii, H. lino, J. Hanna, Appl. Phys. Lett. **2017**, 110, 243301.
- 26) M. Mamada, T. Minamiki, H. Katagiri, S. Tokito, Org. Lett. 2012, 14, 4062.
- 27) J.- II Park, J. W. Chung, J.-Y. Kim, J. Lee, J. Y. Jung, B. Koo, B.-L. Lee, S. W. Lee, Y. W. Jin, S. Y. Lee, J. Am. Chem. Soc. 2015, 137, 12175.
- 28) C. Qian, J. Sun, L. Kong, Y. Fu, Y. Chen, J. Wang, S. Wang, H. Xie, H. Huang, J. Yang, Y. Gao, ACS Photonics **2017**, *4*, 2573.
- J. Sakai, T. Taima, T. Yamanari, K. Saito, Sol. Energy Mater. Sol. Cells 2009, 93, 1149.
- H. Méndez, G. Heimel, S. Winkler, J. Frisch, A. Opitz1, K. Sauer, B. Wegner, M. Oehzelt, C. Röthel, S. Duhm, D. Többens, N. Koch, I. Salzmann, *Nat. Comm.* 2015, *6*, 8560
- M. Ashizawa, T. Niimura, Y. Yu, K. Tsuboi, H. Matsumoto, R. Yamada, S. Kawauchi, A. Tanioka, T. Mori, *Tetrahedron* 2012, 68, 2790.
- S. A. Ponomarenko, S. Kirchmeyer, A Elschner, N. M. Alpatova, M. Halik, H. Klauk, U. Zschieschang, G. Schmid, *Chem. Mater.* 2006, *18*, 579.
- 33) T. J. Dingemans, A. Bacher, M. Thelakkat, L. G. Pedersen, E. T. Samulski, H.-W. Schmidt, Synthetic Metals 1999, 105, 171.
- 34) V. Vohra, G. Arrighetti, L. Barba, K. Higashimine, W. Porzio, H. Murata, J. Phys. Chem. Lett. **2012**, *3*, 1820.
- 35) D. R. T. Zahn, M. Gorgoi, O. D. Gordan, Sol. Ener. 2006, 80, 707.
- 36) W. Tress, K. Leo, M. Riede, Adv. Funct. Mater. 2011, 21, 2140.
- 37) N. Sato, G. Saito, H. Inokuchi, Chem. Phys. 1983, 76, 79.
- 38) H. Yoshida, J. Phys. Chem. C 2014, 118, 24377.
- M. Chikamatsu, A. Itakura, Y. Yoshida, R. Azumi, K. Yase, *Chem. Mater.* 2008, 20, 7365.
- 40) D. R. T. Zahn, G. N. Gavrila. M. Gorgoi, Chem. Phys. 2006, 325, 99.
- 41) B. A. Jones, A. Facchetti, M. R. Wasielewski, T. J. Marks, J. Am. Chem. Soc. 2007, 129, 15259.
- 42) I. Kim, H. M. Haverinen, Z. Wang, S. Madakuni, J. Li, G. E. Jabbour, *Appl. Phys. Lett.* 2009, 95, 023305.
- K. Kanai, K. Akaike, K. Koyasu, K. Sakai, T. Nishi, Y. Kamizuru, T. Nishi, Y. Ouchi, K. Seki, *Appl. Phys. A* **2009**, **95**, 309.
- 44) W. Gao, A. Kahn, Appl. Phys. Lett. 2001, 79, 4040.



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